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Reaction Pathways towards Novel Open Chain and Cyclic Stannasilanes

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Cyclic stannasilaalkanes are formed by rearrangement reactions of open chain and branched Si-Sn derivatives. Furthermore, we report on first attempts towards three membered Si-Sn rings via lithio or potassio substituted stannanes.

Keywords: silicon-tin compounds; stannanes; oligosilanes

INTRODUCTION

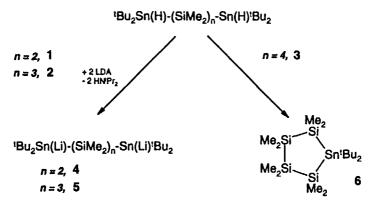
Recently, we reporteded on cyclic Si-Sn compounds^[1, 2] which were synthesized using 'one pot' reactions of oligomeric silanes with chlorostannanes in the presence of magnesium or stepwise syntheses *via* hydrido and chloro substituted Si-Sn chains^[3].

Here we describe base-catalyzed ring closing reactions of linear and branched stannasilane chains yielding unusual five or six membered rings containing tin atoms in addition to silicon.

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RESULTS AND DISCUSSION

As in recent reactions, hydrido-substituted α , ω -stannaoligosilanes (n=2, 1; n=3, 2; n=4, 3) were used as precursors. Reacting 1 with two equivalents of lithium diisopropyl amide (LDA) results in α , ω -dimetallated species 4 and 5 with Si chain lengths of two and three atoms. Surprisingly, in case of the tetrasilane chain the reaction quantitatively yields the stannatetrasilacyclopentane 6.

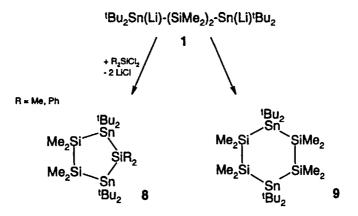


Scheme 1: Reactions of 1-3 with 2 equiv. of LDA

Up to now, it was not possible to clarify the reaction mechanism for the cleavage of the di-tert.butylstannyl group. The α,ω-dimetallated species 5 with three Si atoms in the chain is then reacted with R₂SiCl₂ yielding the expected six-membered ring 7; no byproducts are observed.

$$\begin{array}{c|c} & R_2 \\ & Si \\ & Sn'Bu_2 \\ & & \\ & & \\ & & \\ Me_2Si \\ & & \\ Si \\ & & \\ Si \\ & & \\ & & \\ R = Me, Ph \\ \end{array}$$

When reacting the 1,4-dilithio-substituted stannasilane chain 4 with R₂SiCl₂, we find compound 8 which is accompanied by the unexpected byproduct 9. Seemingly, a reaction similar to the preparation of 6 takes place. In contrast to the four-membered silicon chain, the shorter two-membered chain cannot react on an intramolecular reaction pathway, therefore two Sn-Si-Si fragments combine to compound 9.



Scheme 2: Formation of 8 and 9

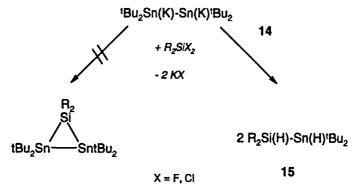
Reactions of the branched stannasilane 10 with three equivalents of LDA yield the cyclopentane 11 and the distannide 12. Similar to the reaction of the open chain tetrasilane, no lithiated Si-Sn products can be observed.^[4]

Scheme 3: Base -catalyzed rearrangement of 10

Treatment of 10 with amines lead to an amine-catalyzed intramolecular ring closing reaction yielding the six membered ring 13. The formation of 13 is the result of a silicon-tin, and surprisingly a tin-carbon bond cleavage; in addition, two tin-tin bonds have been formed.⁽⁴⁾

First attempts towards three-membered Si-Sn rings by reacting the dipotassio distannide 14 with diorganodichloro- or difluorosilanes did not

result in the expected ring systems. Surprisingly, the 1,2-dihydridostannasilanes 15 are formed in nearly quantitative yield.



Scheme 4: Reaction of 14 with diorganodichlorosilanes

Further studies towards the new 1,3-dichloro substituted disilylstannes 16 and 1-sila-2,3-distannapropanes 17 are in progress and will be reported later.

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